Tirucalla-5,24-dien-3 $\beta$-ol [(13 $\alpha, 14 \beta, 17 \alpha, 20 S)$-lanosta-5,24-dien-3 $\beta$ ol $] \dagger$ and three other $\Delta^{5}$-unsaturated tirucallanes from the roots of Bryonia dioica Jacq.: the first naturally occurring C-10 methylated tetracyclic triterpene alcohols with a $\Delta^{5}$-monounsaturated skeleton

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Four novel triterpene alcohols with a $\Delta^{5}$-unsaturated tirucallane-type skeleton, i.e. tirucall-5-en-3p-ol, tirucalla-5,24-dien-3ß-ol, 24-methyltirucalla-5,24(24 ${ }^{1}$-dien-3ß-ol and (24S)-24-methyltirucalla-5,25-dien$3 \beta$-ol, have been isolated from the roots of Bryonia dioica Jacq. (Cucurbitaceae). The structures have been determined by spectroscopic and chromatographic methods. These compounds are the first examples of naturally occurring $\mathbf{C - 1 0}$ methylated triterpenes with a $\Delta^{5}$-monounsaturated skeleton.


#### Abstract

Introduction Bryonia dioica Jacq. (white bryony; Cucurbitaceae) is a climbing perennial herb with tuberous roots native to temperate Europe, North Africa and western Asia. ${ }^{1}$ The roots of B. dioica are characterised by the presence of cucurbitacins, oxygenated tetracyclic triterpenes possessing a wide range of biological activities. ${ }^{2} 3 \beta$-Hydroxy-D : C-friedo-olean-8-en-29-oic acid ( $3 \beta$ -hydroxymultiflor-8-en-29-oic acid; bryonolic acid), ${ }^{3}$ which has been shown to possess a marked anti-allergic activity, ${ }^{4}$ related multiflorane-type triterpenes ${ }^{5}$ and (24E)-5 $\alpha$-stigmasta-$7,24\left(24^{1}\right)$-dien- $3 \beta$-ol (isoavenasterol) ${ }^{6}$ have been isolated from the roots. In our continuing work on the triterpene constituents of Cucurbitaceae, ${ }^{7}$ we now report the isolation and structure elucidation of four novel $\Delta^{5}$-unsaturated tirucallane-type triterpene alcohols from the roots of $B$. dioica.


## Results

The minor and trace components of the triterpene alcohol fraction of $B$. dioica included four novel compounds, viz. tirucall-5-en-3 3 -ol ( $1 \mathbf{a} ; 0.1 \%$ ), tirucalla-5,24-dien-3 $\beta$-ol ( $\mathbf{1 b}$; $1.0 \%$ ), 24-methyltirucalla-5,24(24 ${ }^{1}$ )-dien-3 $\beta$-ol ( $1 \mathrm{c} ; 0.2 \%$ ) and (24S)-24-methyltirucalla-5,25-dien-3 $\beta$-ol ( $1 \mathrm{~d} ; 0.2 \%$ ), and a siructurally related known compound, viz. $5 \alpha$-tirucalla-7,24-dien-3 $\beta$-ol ( $\mathbf{3 b} ; \Delta^{7}$-tirucallol; $1.9 \%$ ), a double-bond isomer of $\mathbf{1 b}$. They were isolated as the acetates ( $2 \mathbf{a}, \mathbf{2 b}, \mathbf{2 c}$ and 2 d ) from the saponified extract of $B$. dioica roots.

Compound 2b $\left[m / z 468\left(\mathrm{M}^{+}, \mathrm{C}_{32} \mathrm{H}_{52} \mathrm{O}_{2}\right)\right]$ had a secondary acetoxy group [ $\delta_{\mathrm{C}} 81.1(\mathrm{~d}) ; \delta_{\mathrm{H}} 2.05(3 \mathrm{H}, \mathrm{s}, \mathrm{OAc})$ and $4.47(1 \mathrm{H}$, dd, $J 7.7$ and 8.1 Hz$)$ ], two trisubstituted double bonds $\left[\delta_{\mathrm{C}}\right.$ $121.4(\mathrm{~d})$ and $125.2(\mathrm{~d}) ; \delta_{\mathrm{H}} 5.54(1 \mathrm{H}, \mathrm{m})$ and $5.10(1 \mathrm{H}, \mathrm{br} t, J 7.0$ $\mathrm{Hz})$ ], a terminal isopropylidene group [ $\delta_{\mathrm{H}} 1.60(\mathrm{~s})$ and $\left.1.68(\mathrm{~s})\right]$, and five tertiary $\left[\delta_{\mathrm{H}} 0.90(6 \mathrm{H}), 0.94\right.$, and $1.01(6 \mathrm{H})$ (each s)]

[^0]and one secondary ( $\delta_{\mathrm{H}} 0.88, \mathrm{~d}, J 6.2 \mathrm{~Hz}$ ) methyl groups. This, in combination with fragment ions having $m / z 453\left(\mathrm{M}^{+}-\mathrm{Me}\right)$, $397\left(\mathrm{M}^{+}-\mathrm{Me}-\mathrm{HOAc}\right), 297$ [loss of side-chain $\left(\mathrm{C}_{8} \mathrm{H}_{15}\right)$ and HOAc], $241\left(297-42-\mathrm{CH}_{2}\right)^{8}$ and $69\left[\mathrm{CH}_{2} \mathrm{CH}=\mathrm{C}(\mathrm{Me})_{2}\right]^{+}$, suggested that compound $\mathbf{2 b}$ was a triterpene with a tetracyclic skeleton possessing one double bond, an equatorially oriented acetoxy group located most likely at $\mathrm{C}-3$, and a $\mathrm{C}_{8}$-side-chain containing an isopropylidene functionality. The highly deshielded $\mathrm{sp}^{2}$ methine ${ }^{1} \mathrm{H}$ multiplet ( $\delta 5.54$ ) due to a skeletal double bond suggested that it was located at C-5(6). ${ }^{9-13}$ Further, the highly deshielded ${ }^{13} \mathrm{C}$ signal at $\delta_{\mathrm{C}} 149.1$ due to $\mathrm{C}-5$ quaternary carbon can be explained by the presence of seven $\beta$-substituted carbons $\left[\left(6 \times \beta^{\sigma}\right)+\left(1 \times \beta^{\pi}\right)\left(\operatorname{sp}^{3}\right)\right]$, ${ }^{14}$ and was consistent with the corresponding signal in the spectrum of $4,4-$ dimethylcholesteryl acetate 5f [ $\left.\delta_{\mathrm{c}} 149.0(\mathrm{C}-5)\right]$. This ruled out the possibility either of a $19(10 \longrightarrow 9)$ abeo- $8 \beta, 9 \beta, 10 \alpha$-lanost-5-ene- (10 $\alpha$-cucurbit-5-ene-) $\left[\begin{array}{lll}\delta_{\mathrm{C}} & 141 & (\mathrm{C}-5)\end{array}\right],{ }^{9 \mathrm{c}} \quad 19(10 \longrightarrow$ 9) abeo-8a, $9 \beta, 10 \alpha$-euph-5-ene- $\left[\begin{array}{llll}\delta_{\mathrm{C}} & 142 & (\mathrm{C}-5)\end{array}\right]^{10}$ or $19(10 \longrightarrow 9)$ abeo- $8 \alpha, 9 \beta, 10 \alpha$-tirucall-5-ene- ${ }^{11}$ skeletal structure. The combined data showed that compound $2 \mathbf{2}$ had a 4,4,14-trimethyl- $\Delta^{5,24}$-cholestadien- $3 \beta$-yl acetate structure with an as-yet-to-be-determined stereochemistry. A weak but diagnostic mass fragment was observed at $m / z 286\left(\mathrm{C}_{21} \mathrm{H}_{34}{ }^{+}\right)$, most likely involving the loss of ring $A$ and part of ring $B$ by cleavages of the C-1-C-10, C-5-C-10 and C-7-C-8 bonds with concomitant ${ }^{1} \mathrm{H}$ loss (we will refer to this fragment ion as $\mathbf{A}$ ) and 271 ( $\mathbf{A}-\mathbf{M e}$ ) supported the $\Delta^{5}$-unsaturation of compound $\mathbf{2 b}$. Fragment $\mathbf{A}$ was observed in the mass spectra of all $\Delta^{5}$ unsaturated tirucallanes and 4,4-dimethylcholesteryl acetate $\mathbf{5 f}$ described in this paper. A similar ion (but formed without ${ }^{1} \mathrm{H}$ transfer) is a diagnostic fragment in the mass spectra of $\Delta^{5}$ unsaturated sterols. ${ }^{15}$ No fragmentation ion due to a retro-Diels-Alder cleavage of ring $\mathbf{B}$, typical of triterpenoids and steroids with $\Delta^{5}$-unsaturation, ${ }^{16}$ was observed in the mass spectra of compound $\mathbf{2 b}$, other $\Delta^{5}$-unsaturated tirucallanes, and $5 f$ described in this paper (see Experimental section). The stereochemistry of compound $\mathbf{2 b}$ was determined by analysis of its 2D NMR data $\left({ }^{1} \mathrm{H}-{ }^{1} \mathrm{H}\right.$ and $\left.{ }^{1} \mathrm{H}-{ }^{13} \mathrm{C}\right)$ and by a nuclear Overhauser enhancement (NOE) study which involved a



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i $(\mathbf{2 0 R}, 24 S)$
comparison of the NOE difference effects in spectra of compound 2 b and in those of four compounds, viz. $\Delta^{7}$-tirucallyl acetate 4b, butyrospermyl acetate $\mathbf{4 e}, 4,4$-dimethylcholest- 5 -en$3 \beta$-yl (4,4-dimethylcholesteryl) acetate 5 f and $5 \alpha$-lanost-7-en$3 \beta$-yl acetate 6 f. NOE Correlations are shown in Fig. 1 .

Compound $\mathbf{5 f}$ showed significant NOE correlation between $\left[28-\mathrm{H}_{3}(4 \alpha-\mathrm{Me}) \sim 3 \alpha-\mathrm{H} \sim 1 \alpha-\mathrm{H} \sim 9 \alpha-\mathrm{H}\right]$ on the $\alpha-$ face and $\left[29-\mathrm{H}_{3}(4 \beta-\mathrm{Me}) \sim 19-\mathrm{H}_{3}(10 \beta-\mathrm{Me}) \sim 8 \beta-\mathrm{H}\right]$ on the $\beta$-face of the molecule. The same significant NOE correlation was observed also for compound 2 b demonstrating that it possessed the same stereochemistry as compound $\mathbf{5 f}$ as far as rings $\mathbf{A}$ and $\mathbf{B}$ and the junction with ring $C$ was concerned. The stereochemistry of the side-chain and of rings C and D was determined by comparison
of the NOE effects in compound $\mathbf{2 b}$, the tirucallane $\mathbf{4 b}$, the euphane 4 e and the lanostane 6 f .

Compound 2b showed NOE correlations between [19$\left.\mathrm{H}_{3} \sim 8 \beta-\mathrm{H} \sim 30-\mathrm{H}_{3}(14 \beta-\mathrm{Me}) \sim 17 \beta-\mathrm{H} \sim 21-\mathrm{H}_{3}\right]$ on the $\beta-$ face, $\left[9 \alpha-\mathrm{H} \sim 18-\mathrm{H}_{3}(13 \alpha-\mathrm{Me}) \sim 20-\mathrm{H}\right]$ on the $\alpha$-face, and $\left[12 \alpha-H \sim 21-\mathrm{H}_{3}\right]$. These NOE correlations were observed also for compound 4b although this exhibited a direct correlation between ( $19-\mathrm{H}_{3} \sim 30-\mathrm{H}_{3}$ ) on the $\beta$-face. The euphane 4 e showed NOE correlations between $\left[19-\mathrm{H}_{3} \sim 30-\mathrm{H}_{3}(14 \beta-\right.$ $\left.\mathrm{Me}) \sim 17 \beta-\mathrm{H} \sim 21-\mathrm{H}_{3}\right]$ on the $\beta-$ face, $\left[9 \alpha-\mathrm{H} \sim 18-\mathrm{H}_{3}\right]$ on the $\alpha$-face, and $\left[16 \alpha, \beta-H \sim 21-\mathrm{H}_{3}\right]$ (Fig. 1). The lanostane $6 f$ showed NOE correlations between $\left[19-\mathrm{H}_{3}(10 \beta-\mathrm{Me}) \sim 18\right.$ $\left.\mathrm{H}_{3}(13 \beta-\mathrm{Me}) \sim 20-\mathrm{H}\right]$ on the $\beta$-face and between $[9 \alpha-\mathrm{H} \sim 30$ $\mathrm{H}_{3}(14 \alpha-\mathrm{Me}) \sim 17 \alpha-\mathrm{H} \sim 21-\mathrm{H}_{3}$ ] on the $\alpha$-face. We concluded that the structure $\mathbf{2 b}$ is that of tirucalla-5,24-dien- $3 \beta$-yl acetate. $\ddagger$
Assigned ${ }^{13} \mathrm{C}$ and ${ }^{1} \mathrm{H}$ NMR data of compounds $\mathbf{1 b}, \mathbf{2 b}, \mathbf{4 b}$ and $\mathbf{4 e}$ are given in Tables 1 and 2, respectively.

Three other novel triterpenes were isolated as the acetates from B. dioica roots: compounds $2 \mathrm{a}\left(\mathrm{m} / \mathrm{z} 470, \mathrm{M}^{+}, \mathrm{C}_{32} \mathrm{H}_{54} \mathrm{O}_{2}\right.$ ), $2 \mathrm{c}\left(\mathrm{m} / \mathrm{z} 482, \mathrm{M}^{+}, \mathrm{C}_{33} \mathrm{H}_{54} \mathrm{O}_{2}\right)$ and $2 \mathrm{~d}\left(\mathrm{~m} / \mathrm{z} 482, \mathrm{M}^{+}, \mathrm{C}_{33} \mathrm{H}_{54} \mathrm{O}_{2}\right)$. The ${ }^{1} \mathrm{H}$ NMR spectra of these acetates and the corresponding free alcohols included signals of the ring system similar to those of compounds $\mathbf{2 b} / \mathbf{1 b}$ (see Tables 2 and 3). This suggested a tirucall- or an euph-5-en-3 3 -ol structure.§ The structures of the side-chains were determined by comparison of the ${ }^{1} \mathrm{H}$ NMR data with those of related compounds in the literature. ${ }^{19}$
The HPLC and GLC retention factors ( $R_{\mathrm{f}}$ ) of four triterpenes $\mathbf{2 a}, \mathbf{2 b}, \mathbf{2 c}$ and 2d, calculated from their relative retention times ( $t_{\mathrm{R}}$ ), were in excellent agreement with those of a set of cycloartanes, which have the same side-chains but the opposite configuration at C-20, viz. $5 \alpha$-cycloartan- $3 \beta$-yl acetate 7 f , $5 \alpha-$ cycloart-24-en-3 3 -yl acetate $7 \mathrm{~g}, 24$-methyl- $5 \alpha$-cycloart-24( $24^{1}$ )-en- $3 \beta$-yl acetate 7 h and ( 24,5 )-24-methyl- $5 \alpha$-cycloart- 25 -en- $3 \beta$ yl acetate 7 i , respectively (see Table 4). This showed that all four novel compounds were tirucallanes.

## Discussion

Several naturally occurring tetracyclic ${ }^{9-11}$ and pentacyclic triterpenes ${ }^{12,13,20}$ with a C-9 methylated [19(10 $\left.\longrightarrow 9\right)$ abeo] $\Delta^{5}$-unsaturated skeleton have been reported. However, the four tirucallane-type triterpenes $1 \mathbf{1 a - d}$ are the first examples of triterpene alcohols with a C - 10 methylated $\Delta^{5}$-monounsaturated skeleton. $\dagger \mathrm{C}$-10 Methylated triterpenes with a
$\ddagger$ The most stable conformation of $\mathbf{2 b}$ ( $66.44 \mathrm{kcal} \mathrm{mol}^{-1} ; 1 \mathrm{cal}=4.184$ J) with minimum steric energy was simulated by using CAChe and MM2 programs ${ }^{17}$ and is shown in Fig. 1. The same simulation was also carried out for compounds 4 b ( $61.03 \mathrm{kcal} \mathrm{mol}^{-1}$ ), $\mathbf{4 e}\left(60.72 \mathrm{kcal} \mathrm{mol}^{-1}\right.$ ) and $5 \mathrm{f}\left(55.19 \mathrm{kcal} \mathrm{mol}^{-1}\right)$, and the results are also shown in Fig. 1. The simulated most stable conformer of compound $\mathbf{2 b}$ orients $\mathbf{C}-22$ in a 'right-handed' conformation (C-22 trans-oriented with respect to $\mathrm{C}-13$ ) similar to that of compound 4 b and to the crystal structure of another tirucallane, $5 \alpha$-tirucalla- 8,24 -dien- $3 \beta$-yl (tirucallyl) acetate. ${ }^{18}$ This conformation of compound $\mathbf{2 b}$ was fairly consistent with results from the NOE experiment carried out in solution. In the simulated most stable conformation of compound $4 \mathrm{e}, \mathrm{C}-22$ was cis-oriented ('lefthanded') with respect to C-13, which was consistent with the NOE experimental results and with the crystal structure of $5 \alpha$-eupha-8,24-dien-3 $\beta$-yl (euphyl) acetate. ${ }^{18}$
$\S{ }^{1} \mathrm{H}$ and ${ }^{13} \mathrm{C}$ NMR data are almost useless for distinguishing between euphane- and tirucallane-type triterpenes. Examples are the NMR data of $\mathbf{4 b}$, a tirucallane, and $\mathbf{4 e}$, a euphane, which are very similar (Tables 1 and 2 ).

- The occurrence of two triterpenes with a C - 10 methylated $\Delta^{5}$-monounsaturated skeleton, viz. lupa-5,20(29)-dien-3 $\beta$-ol in the bark of Holarrhena antidysenterica ${ }^{21}$ and lupa-5,20(29)-en-3-one in the stem bark of Pleurostylia opposita has been reported. ${ }^{22}$ The structural assignments of these triterpenes should be reinvestigated because they were reported to give a C-6 vinyl proton resonance at somewhat higher field, viz. $\delta 5.08^{21}$ and $5.40,{ }^{22}$ respectively, in the ${ }^{1} \mathrm{H}$ NMR $\left(\mathrm{CDCl}_{3}\right)$


Fig. 1 CAChe drawings and some representative NOE correlations (-) for compounds $\mathbf{2 b}, \mathbf{4 b}, \mathbf{4 e}$ and $\mathbf{5 f}$
Table $1{ }^{13} \mathrm{C}$ NMR spectral data ( $\delta$ values; $100.62 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) of some tirucallane- and euphane-type triterpenes with a $\Delta^{24}$-unsaturated sidechain isolated from B. dioica roots

| Carbon | 1b | 2b | 4b | 4e | Carbon | 1b | 2b | 4b | 4 e |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 35.5 | 35.2 | 36.8 | 36.8 | 17 | 53.6 | 53.5 | 52.9 | 53.2 |
| 2 | 28.0 | 24.3 | 24.2 | 24.2 | 18 | 23.6 | 23.6 | 21.9 | 22.1 |
| 3 | 79.9 | 81.1 | 81.1 | 81.1 | 19 | 24.5 | 24.5 | 13.2 | 13.1 |
| 4 | 38.8 | 37.7 | 37.8 | 37.8 | 20 | 36.1 | 36.0 | 35.9 | 35.8 |
| 5 | 149.0 | 149.1 | 50.8 | 50.8 | 21 | 18.3 | 18.3 | 18.3 | 18.6 |
| 6 | 121.4 | 121.1 | 23.8 | 23.8 | 22 | 36.2 | 36.2 | 36.2 | 35.2 |
| 7 | 23.1 | 22.9 | 117.6 | 117.6 | 23 | 25.1 | 25.1 | 25.0 | 25.4 |
| 8 | 48.6 | 48.6 | 146.0 | 146.0 | 24 | 125.2 | 125.2 | 125.2 | 125.1 |
| 9 | 48.4 | 48.4 | 48.9 | 48.8 | 25 | 131.0 | 131.0 | 130.9 | 131.0 |
| 10 | 35.9 | 35.7 | 34.8 | 34.8 | 26 | 25.7 | 25.2 | 25.7 | 25.7 |
| 11 | 23.0 | 22.9 | 18.1 | 18.1 | 27 | 17.7 | 17.7 | 17.7 | 17.7 |
| 12 | 35.3 | 35.2 | 33.7 | 33.7 | 28 | 28.9 | 28.8 | 27.6 | 27.6 |
| 13 | 43.5 | 43.5 | 43.5 | 43.5 | 29 | 16.4 | 17.5 | 15.9 | 15.9 |
| 14 | 52.8 | 52.8 | 51.2 | 51.3 | 30 | 30.5 | 30.5 | 27.3 | 27.3 |
| 15 | 33.5 | 33.4 | 34.0 | 33.9 | COMe (3') |  | 171.0 | 171.0 | 171.0 |
| 16 | 28.7 | 28.6 | 28.2 | 28.5 | COMe (3") |  | 21.3 | 21.3 | 21.3 |

$\Delta^{5}$-bond and an additional double bond in the skeleton are known constituents of the seeds of two members of the Cucurbitaceae. D:C-friedo-Oleana-5,7,9(11)-triene-3 2 ,29-diol (5-dehydrokarounidiol) and 11-oxolanosta-5,16,20,25-tetraen$3 \beta$-ol (citrullonol) occur in the seeds of Trichosanthes kirilowii ${ }^{7 c}$ and of Citrullus colocynthis, ${ }^{23}$ respectively.

It is tempting to speculate that compound $\mathbf{1 b}$, the logical precursor of compounds $\mathbf{1 a}, \mathbf{1 c}$ and $\mathbf{1 d},{ }^{24}$ is formed by cyclis-
spectrum. $\Delta^{5}$-Unsaturated $3 \beta$-hydroxy (and acetoxy) triterpenes afford the C-6 vinyl proton at $\delta \sim 5.5-5.6$ as shown in the literature ${ }^{9-13}$ and in the present study. Moreover, the C-6 vinyl proton of 4,4-dimethylcholest-5-en-3-one was observed at $\delta 5.55$ (dd, $J 2.2$ and 5.1 Hz ) in the ${ }^{1} \mathrm{H}$ NMR ( $\mathrm{CDCl}_{3} ; 400 \mathrm{MHz}$ ) spectrum (unpublished results).
ation of squalene 2,3-oxide (to give the protoeuphoid cation 8 ) followed by a series of 1,2 -shifts and loss of one hydrogen (see Scheme 1). \| An alternative route might involve a C-9 carbocation and a 1,3-transannular hydrogen shift from C-5.

## Experimental

## General

Crystallisations were performed from methanol. Mps were measured on a Yanagimoto micro mp apparatus and are
|| It has been suggested that a series of 1,2 -shifts in the protoeuphoid cation can also result in formation of the $19(10 \rightarrow 9)$ abeo-euph- 5 -ene skeleton. ${ }^{10}$

Table $2{ }^{1} \mathrm{H}$ NMR Spectral data ( $\delta$ values; $400 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) of some tirucallane- and euphane-type triterpenes with a $\Delta^{24}$-unsaturated side-chain isolated from B. dioica roots ${ }^{a}$

| Proton | 1b | 2b | 4d | 4e |
| :---: | :---: | :---: | :---: | :---: |
| $1-\mathrm{H}_{2}$ | 1.42 ( 2 H ) | $1.43(\alpha), 1.58(\beta)$ | $1.26(\alpha), 1.66(\beta)$ | 1.22( $\alpha$ ), 1.66( $\beta$ ) |
| $2-\mathrm{H}_{2}$ | 1.64 (2 H) | 1.50( $\alpha$ ), 1.66( $\beta$ ) | 1.67 (2 H) | 1.67 (2 H) |
| $3 \alpha-\mathrm{H}$ | 3.21 (dd, 7.6, 7.6) | 4.47 (dd, 7.7, 8.1) | 4.52 (dd, 4.8, 11.4) | 4.52 (dd, 4.0, 11.0) |
| $5 \alpha-\mathrm{H}$ |  |  | 1.41 | 1.41 (dd, 5.9, 12.1) |
| 6-H | 5.56 (ddd, 2.8, 2.8, 3.6) | 5.54 | $2.13(\alpha), 1.93(\beta)$ | $2.13(\alpha), 1.96$ ( $\beta$ ) |
| 7-H | 1.92 (2 H) | 1.92 (2 H) | 5.25 (dd, 2.9, 7.0) | 5.25 (dd, 2.7, 6.6) |
| $8 \beta-\mathrm{H}$ | 0.85 | 0.92 |  |  |
| $9 \alpha-\mathrm{H}$ | 2.27 (br d, 14.8) | 2.28 (br d, 15.0) | 2.23 | 2.22 |
| $11-\mathrm{H}_{2}$ | $1.70(\alpha), 1.43(\beta)$ | $1.70(\alpha), 1.44(\beta)$ | 1.52 (2 H) | 1.52 (2 H) |
| $12-\mathrm{H}_{2}$ | $1.66(\alpha), 1.81(\beta)$ | $1.63(\alpha), 1.80(\beta)$ | $1.64(\alpha), 1.78(\beta)$ | 1.66( $\alpha$ ), 1.80( $\beta$ ) |
| $15-\mathrm{H}_{2}$ | $1.50(2 \mathrm{H})$ | $1.51(2 \mathrm{H})$ | $1.53(2 \mathrm{H})$ | $1.45(2 \mathrm{H})$ |
| 16-H2 | $1.28(\alpha), 1.96(\beta)$ | $1.26(\alpha), 1.94(\beta)$ | $1.30(\alpha), 1.96(\beta)$ | $1.27(\alpha), 1.92(\beta)$ |
| 178-H | 1.46 | 1.48 | 1.48 | 1.49 |
| $18-\mathrm{H}_{3}$ | 0.90 (s) | 0.90 (s) | 0.81 (s) | 0.80 (s) |
| $19-\mathrm{H}_{3}$ | 0.98 (s) | 1.01 (s) | 0.77 (s) | 0.77 (s) |
| 20-H | 1.38 | 1.39 | 1.42 | 1.40 |
| $21-\mathrm{H}_{3}$ | 0.88 (d, 5.6) | 0.88 (d, 6.2) | 0.88 (d, 6.2) | 0.85 (d, 6.2) |
| $22-\mathrm{H}_{2}$ | 1.04, 1.44 | 1.04, 1.45 | 1.03, 1.43 | 0.99, 1.59 |
| $23-\mathrm{H}_{2}$ | 1.90, 2.04 | 1.88, 2.02 | 1.87, 2.04 | 1.88, 2.04 |
| 24-H | 5.10 (br t, 7.2) | 5.10 (br t, 7.0) | 5.10 (br t, 7.3) | 5.10 (br t, 7.0) |
| $26-\mathrm{H}_{3}$ | 1.68 (s) | 1.68 (s) | 1.69 (s) | 1.69 (s) |
| $27-\mathrm{H}_{3}$ | 1.60 (s) | 1.60 (s) | 1.61 (s) | 1.61 (s) |
| $28-\mathrm{H}_{3}$ | 1.02 (s) | 0.90 (s) | 0.85 (s) | 0.85 (s) |
| $29-\mathrm{H}_{3}$ | 0.87 (s) | 0.94 (s) | 0.93 (s) | 0.93 (s) |
| $30-\mathrm{H}_{3}$ | 1.02 (s) | 1.01 (s) | 0.97 (s) | 0.97 (s) |
| $3 \beta-\mathrm{OAc}$ |  | 2.05 (s) | 2.06 (s) | 2.05 (s) |

${ }^{a} J$ Values $(\mathrm{Hz})$ are bracketed. $J$ Values not included in the Table were not determined.
Table $3{ }^{1} \mathrm{H}$ NMR Spectral data ( $\delta$ values; $400 \mathrm{MHz} ; \mathrm{CDCl}_{3}$ ) of three novel tirucallane-type triterpene alcohols and their acetates reported in this paper ${ }^{\text {a }}$

| Proton | 1 a | 2a | 1c | 2c | 1d | 2d |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| $3 \alpha-\mathrm{H}$ | 3.21 | 4.47 (dd-like, 7.7, 8.1) | 3.21 (dd-like, 5.9, 9.8) | 4.47 (dd-like, 6.0, 8.4) | 3.21 | 4.47 (dd-like, 7.7, 8.1) |
| 6-H | 5.57 | 5.54 | 5.56 (ddd, 3.0, 3.0, 6.9) | 5.55 | 5.55 (ddd, 2.8, 2.8, 6.6) | 5.54 |
| $18-\mathrm{H}_{3}$ | 0.90 (s) | 0.90 (s) | 0.90 (s) | 0.90 (s) | 0.89 (s) | 0.89 (s) |
| $19-\mathrm{H}_{3}$ | 0.99 (s) | 0.99 (s) | 0.99 (s) | 1.01 (s) | 0.98 (s) | 1.01 (s) |
| $21-\mathrm{H}_{3}$ | 0.86 (d, 6.4) | 0.86 (d, 6.4) | 0.89 (d, 6.3) | 0.89 (d, 6.6) | 0.85 (d, 6.3) | 0.85 (d, 6.3) |
| 25-H | n.d. | n.d. | 2.24 (sept., 7.4) | 2.23 (sept., 6.8) |  |  |
| $26-\mathrm{H}_{3}$ | 0.86 (d, ${ }^{\text {b }} 6.0$ ) | 0.86 (d, ${ }^{\text {b }} 6.0$ ) | 1.03 (d, ${ }^{\text {b }} 6.6$ ) | 1.03 (d, 6.9) | 1.64 (s) | 1.64 (s) |
| 27-H | 0.87 (d, ${ }^{\text {b }} 6.4$ ) | 0.87 (d, ${ }^{\text {b }} 6.4$ ) | 1.03 (d, ${ }^{\text {b }} 6.9$ ) | 1.03 (d, 6.9) | 4.67 ( $2 \mathrm{H}, \mathrm{br} \mathrm{s}$ ) | 4.67 ( $2 \mathrm{H}, \mathrm{br} \mathrm{s}$ ) |
| $24^{1}-\mathrm{H}$ |  |  | 4.66 (d, 1.4) | 4.66 (d, 1.4) | 1.00 (d, 6.9) | 1.00 (d, 7.2) |
|  |  |  | 4.72 (s) | 4.72 (s) |  |  |
| $28-\mathrm{H}_{3}$ | 1.02 (s) | 0.90 (s) | 1.02 (s) | 0.90 (s) | 1.02 (s) | 0.90 (s) |
| $29-\mathrm{H}_{3}$ | 0.87 (s) | 0.94 (s) | 0.87 (s) | 0.94 (s) | 0.87 (s) | 0.94 (s) |
| $30-\mathrm{H}_{3}$ | 1.02 (s) | 1.02 (s) | 1.02 (s) | 1.03 (s) | 1.02 (s) | 1.01 (s) |
| 3及-OAc |  | 2.05 (s) |  | 2.05 (s) |  | 2.05 (s) |

${ }^{a} J$ Values $(\mathrm{Hz})$ are bracketed. $J$ Values not included in the Table were not determined. ${ }^{b}$ Assignments in each column are interchangeable. n.d. $=$ not determined.

Table 4 Relative retention times $\left(\mathrm{R} t_{\mathrm{R}}\right)^{a}$ and retention factor $\left(R_{\mathrm{f}}\right)^{b}$ of the acetyl derivatives of some triterpene alcohols from $B$. dioica roots, and of cycloartane triterpene alcohols

| Triterpene acetate | GLC |  | HPLC |  |
| :---: | :---: | :---: | :---: | :---: |
|  | $\mathrm{R} t_{\mathrm{R}}(\mathrm{I})$ | $R_{\text {f }}$ | $\mathrm{R} t_{\mathrm{R}}(\mathrm{I})$ | $R_{\text {f }}$ |
| Cycloartane group |  |  |  |  |
| $5 \alpha$-Cycloartan-3 $\beta$-ol 7 f (cycloartanol) | 1.50 | 1.00 | 1.26 | 1.00 |
| $5 \alpha$-Cycloart-24-en-3 3 -ol 7g (cycloartenol) | 1.82 | 1.21 | 1.01 | 0.80 |
| 24-Methyl-5 $\alpha$-cycloart-24(24 ${ }^{1}$ )-en-3 $\beta$-ol 7h (24-methylenecycloartanol) | 2.00 | 1.33 | 1.09 | 0.87 |
| (24S)-24-Methyl-5 $\alpha$-cycloart-25-en-3 $\beta$-ol 7 i (cyclolaudenol) | 1.95 | 1.30 | 1.08 | 0.86 |
| $\Delta^{5}$-Unsaturated tirucallane group |  |  |  |  |
| Tirucall-5-en-3 $\beta$-ol 2 a | 1.35 | 1.00 | 0.92 | 1.00 |
| Tirucalla-5,24-dien-3 $\beta$-ol 2b | 1.64 | 1.21 | 0.74 | 0.80 |
| 24-Methyltirucalla-5,24(24 ${ }^{1}$ )-dien-3 $\beta$-ol 2c | 1.81 | 1.34 | 0.80 | 0.87 |
| (24S)-24-Methyltirucalla-5,25-dien-3 $\beta$-ol 2d | 1.77 | 1.31 | 0.79 | 0.86 |

[^1]uncorrected. Argentic TLC plates [silica gel- $\mathrm{AgNO}_{3}$ (4:1)] were developed twice with $\mathrm{CCl}_{4}-\mathrm{CH}_{2} \mathrm{Cl}_{2}$ (4:1). HPLC Separations were performed using an Ultrasphere ODS column
( $5 \mu ; 25 \mathrm{~cm} \times 10 \mathrm{~mm}$ i.d., Beckman Instruments, Inc., California) using MeOH at $4 \mathrm{ml} \mathrm{min}^{-1}$ and a refractive-index detector. A DB-17 fused silica capillary column ( $30 \mathrm{~m} \times 0.3$



Scheme 1 Possible biosynthetic route for the formation of $\Delta^{5}$ unsaturated tirucallane triterpene alcohols (1a-d)
mm i.d.; $275^{\circ} \mathrm{C}$ ) was used for GLC. In both HPLC and GLC, cholesteryl (cholest-5-en- $3 \beta$-yl) acetate was the standard for the determination of $\mathrm{R} t_{\mathrm{R}}(\mathrm{I})$ of acetoxy triterpenes; cholesterol was the standard for the determination of $\mathbf{R} t_{\mathbf{R}}$ (II) for the hydroxy triterpenes. EI-MS were recorded on a Hitachi M-80B double-focussing GC-MS instrument ( 70 eV ) using a direct inlet system. NMR Spectra were recorded with JEOL GX-400 and GSX-400 spectrometers at $400 \mathrm{MHz}\left({ }^{1} \mathrm{H}\right.$ NMR) and 100.62 $\mathrm{MHz}\left({ }^{13} \mathrm{C} \mathrm{NMR}\right)$ in $\mathrm{CDCl}_{3}$ with $\mathrm{Me}_{4} \mathrm{Si}\left({ }^{1} \mathrm{H}\right.$ NMR) and $\mathrm{CDCl}_{3}$ at $\delta_{\mathrm{C}} 77.0\left({ }^{13} \mathrm{C}\right.$ NMR) as internal standard. $J$ Values are given in Hz . Acetylation was performed in $\mathrm{Ac}_{2} \mathrm{O}$-pyridine at room temperature overnight, whereas acetates were hydrolysed in $5 \%$ KOH in MeOH at room temperature overnight. The following triterpene acetates were used as reference compounds: $\mathbf{4 b}, \mathbf{4 e}, 7 \mathrm{f}$, $\mathbf{7 g}, \mathbf{7 h}, 7 \mathbf{7 i},{ }^{25} \mathbf{5 f}^{26}$ and $\mathbf{6 f}$. ${ }^{27}$ The roots of B. dioica Jacq. were collected in the Netherlands in late September, 1986.

## Isolation procedure

Air-dried and ground roots of B. dioica ( 12.5 kg ) were extracted with hexane and then with MeOH under reflux. Neutral lipids $(5.9 \mathrm{~g})$ were obtained from the combined extracts $(500 \mathrm{~g})$ by alkaline hydrolysis ( $5 \% \mathrm{KOH}$ in MeOH ; reflux; 3 h ). The neutral lipids were chromatographed over silica gel ( 250 g ) with hexane, hexane-EtOAc ( $9: 1, \mathrm{v} / \mathrm{v}$ ) and hexane-EtOAc (4:1) as eluents. The residue of the hexane- $\operatorname{EtOAc}(9: 1)$ eluate yielded a triterpene alcohol fraction ( 376 mg ) after rechromatography over silica gel. The fraction was acetylated, and the resulting acetate fraction ( 374 mg ) was subjected to argentic TLC followed by HPLC. The following triterpene acetates discussed in this paper were obtained: $\mathbf{2 a}(0.2 \mathrm{mg}), \mathbf{2 b}(3.6 \mathrm{mg}), \mathbf{2 c}(0.7$ $\mathrm{mg}), \mathbf{2 d}(0.5 \mathrm{mg}), \mathbf{4 b}(7.0 \mathrm{mg}), \mathbf{4 e}(11 \mathrm{mg}), 7 \mathrm{~g}(55 \mathrm{mg})$ and $7 \mathrm{~h}(18$ mg ). Known triterpenes ( $\mathbf{4 b}, \mathbf{4 e}, 7 \mathrm{~g}$ and $\mathbf{7 h}$ ) were identified by chromatographic (HPLC, GLC) and spectral (mass, ${ }^{1} \mathrm{H}$ NMR) comparison with reference compounds. All identified triterpene alcohols, $4 \alpha$-methyl sterols and sterols isolated from B. dioica roots have been reported in another paper. ${ }^{28}$

Tirucall-5-en-3及-yl acetate 2 a and tirucall-5-en-3 $\beta$-ol 1a.

Compound 2a: $m / z$ (assignment) $470.4111\left(\mathrm{C}_{32} \mathrm{H}_{54} \mathrm{O}_{2}, \mathrm{M}^{+}\right.$, $6 \%$; requires $M, 470.4120$ ), $455.3889\left(\mathrm{C}_{31} \mathrm{H}_{51} \mathrm{O}_{2}, 26\right), 395.3628$ $\left(\mathrm{C}_{29} \mathrm{H}_{27}, 18\right), 315.2382\left(\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{O}_{2}, 1\right), 297.2569\left(\mathrm{C}_{22} \mathrm{H}_{33}, 1\right)$, $288.2797\left(\mathrm{C}_{21} \mathrm{H}_{36}, A, 3\right), 283.2438\left(\mathrm{C}_{21} \mathrm{H}_{31}, 1\right), 273.2543$ $\left(\mathrm{C}_{22} \mathrm{H}_{33}, \mathrm{~A}-\mathrm{Me}, 5\right), 241.1985\left(\mathrm{C}_{18} \mathrm{H}_{25}, 2\right), 229.1991\left(\mathrm{C}_{17} \mathrm{H}_{25}\right.$, 5), $43.0578\left(\mathrm{C}_{3} \mathrm{H}_{7}\right)$ and $43.0217\left(\mathrm{C}_{2} \mathrm{H}_{3} \mathrm{O}, 100\right)$. Alkaline hydrolysis of compound $\mathbf{2 a}$ afforded the alcohol $1 \mathrm{a}, \mathrm{R} t_{\mathrm{R}}(\mathrm{II})$ (GLC) $1.48 ; m / z 428.3987\left(\mathrm{C}_{30} \mathrm{H}_{52} \mathrm{O}, \mathrm{M}^{+}, 2 \%\right.$; requires $M$, 428.4015), $413.3819\left(\mathrm{C}_{29} \mathrm{H}_{49} \mathrm{O}, 5\right), 395.3695\left(\mathrm{C}_{29} \mathrm{H}_{47}, 3\right)$, $288.2798\left(\mathrm{C}_{21} \mathrm{H}_{36}, A, 3\right), 273.2518\left(\mathrm{C}_{20} \mathrm{H}_{33}, \mathrm{~A}-\mathrm{Me}, 6\right)$, $259.2090\left(\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{O}, 3\right), 255.2138\left(\mathrm{C}_{19} \mathrm{H}_{27}, 5\right), 241.1980$ $\left(\mathrm{C}_{18} \mathrm{H}_{25}, 2\right), 229.2011\left(\mathrm{C}_{17} \mathrm{H}_{25}, 4\right)$ and $43.0542\left(\mathrm{C}_{3} \mathrm{H}_{7}, 100\right)$.

Tirucalla-5,24-dien- $\mathbf{3 \beta}$-yl acetate $\mathbf{2 b}$ and tirucalla-5,24-dien-3p-ol 1b. Compound 2b: mp $138-140^{\circ} \mathrm{C} ; \mathrm{m} / \mathrm{z} 468.3938$ $\left(\mathrm{C}_{32} \mathrm{H}_{52} \mathrm{O}_{2}, \mathrm{M}^{+}, 24 \%\right.$; requires $\left.M, 468.3964\right), 453.3689$ $\left(\mathrm{C}_{31} \mathrm{H}_{49} \mathrm{O}_{2}, 47\right), 408.3629\left(\mathrm{C}_{30} \mathrm{H}_{48}, 2\right), 393.3477\left(\mathrm{C}_{29} \mathrm{H}_{45}, 37\right)$, $315.2361\left(\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{O}_{2}, 2\right), 297.2555\left(\mathrm{C}_{22} \mathrm{H}_{33}, 3\right), 295.2407$ $\left(\mathrm{C}_{22} \mathrm{H}_{31}, 1\right), 286.2665\left(\mathrm{C}_{21} \mathrm{H}_{34}, A, 3\right), 271.2424\left(\mathrm{C}_{20} \mathrm{H}_{31}, \mathbf{A}-\right.$ $\mathrm{Me}, 9), 257.2244\left(\mathrm{C}_{19} \mathrm{H}_{29}, 4\right), 255.2062\left(\mathrm{C}_{19} \mathrm{H}_{27}, 5\right), 241.1907$ $\left(\mathrm{C}_{18} \mathrm{H}_{25}, 7\right)$ and $69.0695\left(\mathrm{C}_{5} \mathrm{H}_{9}, 100\right)$. Alkaline hydrolysis of acetate $\mathbf{2 b}$ yielded the alcohol $1 \mathbf{b}, \mathrm{mp}, 139-140^{\circ} \mathrm{C}, \mathbf{R} t_{\mathbf{R}}(\mathrm{II})$ (GLC) $1.78 ; \mathrm{m} / \mathrm{z} 426.3847\left(\mathrm{C}_{30} \mathrm{H}_{50} \mathrm{O}, \mathrm{M}^{+}, 13 \%\right.$; requires $M$, 426.3859), $411.3610\left(\mathrm{C}_{29} \mathrm{H}_{47} \mathrm{O}, 25\right)$, $393.3508\left(\mathrm{C}_{29} \mathrm{H}_{45}, 7\right)$, $341.2787\left(\mathrm{C}_{24} \mathrm{H}_{37} \mathrm{O}, 1\right), 286.2673\left(\mathrm{C}_{21} \mathrm{H}_{34}, 3\right), 271.2413$ $\left(\mathrm{C}_{20} \mathrm{H}_{31}, \mathrm{~A}-\mathrm{Me}, 4\right), 259.2122\left(\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{O}, 2\right), 255.2062$ $\left(\mathrm{C}_{19} \mathrm{H}_{27}, 1\right), 243.2087\left(\mathrm{C}_{18} \mathrm{H}_{27}, 2\right), 241.1924\left(\mathrm{C}_{18} \mathrm{H}_{25}, 2\right)$, $229.1990\left(\mathrm{C}_{17} \mathrm{H}_{25}, 2\right)$, $215.1828\left(\mathrm{C}_{16} \mathrm{H}_{25}, 3\right)$, $201.1670\left(\mathrm{C}_{15} \mathrm{H}_{21}\right.$, 5) and $69.0706\left(\mathrm{C}_{5} \mathrm{H}_{9}, 100\right)$.

The NOE correlations for acetate $\mathbf{2 b}$ shown in Fig. 1 were determined using the difference NOE spectral technique. The representative correlations observed were as follows. Irradiation of the signal at $\delta 4.47(3 \alpha-\mathrm{H})$ enhanced signals at $\delta 0.90\left[28-\mathrm{H}_{3}\right.$ $(4 \alpha-\mathrm{Me})]$ and $1.43(1 \alpha-\mathrm{H})$. Irradiation of the signal at $\delta 1.01$ [19$\mathrm{H}_{3}$ and $\left.30-\mathrm{H}_{3}(14 \beta-\mathrm{Me})\right]$ enhanced signals at $\delta 0.92(8 \beta-\mathrm{H}), 0.94$ $\left[29-\mathrm{H}_{3}(4 \beta-\mathrm{Me})\right]$ and $1.48(17 \beta-\mathrm{H})$, whereas irradiation at $\delta$ $0.92(8 \beta-\mathrm{H})$ enhanced a signal at $\delta 1.01\left(19-\mathrm{H}_{3}\right.$ and $\left.30-\mathrm{H}_{3}\right)$. Irradiation at $\delta 1.48(17 \beta-\mathrm{H})$ enhanced a signal at $\delta 1.01$ $\left(30-\mathrm{H}_{3}\right)$. Irradiation at $\delta 0.88\left(21-\mathrm{H}_{3}\right)$ enhanced signals at $\delta 1.39$ $(20-\mathrm{H}), 1.48(17 \beta-\mathrm{H})$ and $1.63(12 \beta-\mathrm{H})$. Irradiation at $\delta 2.28$ $(9 \alpha-\mathrm{H})$ enhanced a signal at $\delta 0.90\left[18-\mathrm{H}_{3}(13 \alpha-\mathrm{Me})\right]$. Finally, irradiation of the signal at $\delta 0.90\left(18-\mathrm{H}_{3}\right.$ and $\left.28-\mathrm{H}_{3}\right)$ enhanced signals at $\delta 1.39(20-\mathrm{H})$ and $4.47(3 \alpha-\mathrm{H})$. The presence of two overlapped methyl signals at $\delta 0.90\left(18-\mathrm{H}_{3}\right.$ and $\left.28-\mathrm{H}_{3}\right)$ and 1.01 $\left(19-\mathrm{H}_{3}\right.$ and $30-\mathrm{H}_{3}$ ) in the NMR spectrum of acetate 2 b caused some ambiguity in the assignment of NOE correlations, which was overcome by comparison of the NOE correlations observed for free alcohol 1b. The representative NOE correlations for alcohol $\mathbf{1 b}$ were as follows. Irradiation of the signal at $\delta 3.21$ $(3 \alpha-\mathrm{H})$ enhanced signals at $\delta 1.02\left(28-\mathrm{H}_{3}\right)$ and $1.42(1 \alpha-\mathrm{H})$. Irradiation at $\delta 0.98\left(19-\mathrm{H}_{3}\right)$ enhanced signals at $\delta 0.85(8 \beta-\mathrm{H})$ and $0.87\left(29-\mathrm{H}_{3}\right)$, whereas irradiation at $\delta 0.85(8 \beta-\mathrm{H})$ enhanced signals at $\delta 0.98\left(19-\mathrm{H}_{3}\right)$ and $1.02\left(30-\mathrm{H}_{3}\right)$. Further irradiation at $\delta 1.02\left(28-\mathrm{H}_{3}\right.$ and $\left.30-\mathrm{H}_{3}\right)$ enhanced signals at $\delta 0.85(8 \beta-\mathrm{H}), 0.87$ $\left(29-\mathrm{H}_{3}\right), 1.46(17 \beta-\mathrm{H})$ and $3.21(3 \alpha-\mathrm{H})$, while irradiation at $\delta$ $1.46(17 \beta-\mathrm{H})$ enhanced a signal at $\delta 1.02\left(30-\mathrm{H}_{3}\right)$. Irradiation at $\delta 0.88\left(21-\mathrm{H}_{3}\right)$ enhanced signals at $\delta 1.38(20-\mathrm{H})$ and $1.66(12 \beta-$ H). Irradiation at $\delta 0.90\left(18-\mathrm{H}_{3}\right)$ enhanced signals at $\delta 1.38(20-$ H) and $2.27(9 \alpha-\mathrm{H})$, whereas irradiation at $\delta 2.27(9 \alpha-\mathrm{H})$ enhanced signals at $\delta 0.90\left(18-\mathrm{H}_{3}\right)$ and $1.42(1 \alpha-\mathrm{H})$.

24-Methyltirucalla-5,24(241)-dien-3p-yl acetate 2c and 24-methyltirucalla-5,24(24 ${ }^{1}$ )-dien-3p-ol 1c. Compound 2c: $m / z$ $482.4123\left(\mathrm{C}_{33} \mathrm{H}_{54} \mathrm{O}_{2}, \mathrm{M}^{+}, 9 \%\right.$; requires $M, 482.4121$ ), 467.3880 $\left(\mathrm{C}_{32} \mathrm{H}_{51} \mathrm{O}_{2}, 21\right), 422.3841\left(\mathrm{C}_{31} \mathrm{H}_{50}, 1\right), 407.3651\left(\mathrm{C}_{30} \mathrm{H}_{47}, 7\right)$, $383.2941\left(\mathrm{C}_{26} \mathrm{H}_{39} \mathrm{O}_{2}, 1\right)$, $369.2778\left(\mathrm{C}_{25} \mathrm{H}_{37} \mathrm{O}_{2}, 1\right.$ ), 355.2671 $\left(\mathrm{C}_{24} \mathrm{H}_{35} \mathrm{O}_{2}, 1\right), 323.2740\left(\mathrm{C}_{24} \mathrm{H}_{35}, 2\right), 315.2273\left(\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{O}_{2}, 1\right)$, $301.2173\left(\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{O}_{2}, 4\right), 300.2792\left(\mathrm{C}_{22} \mathrm{H}_{36}, \mathrm{~A}, 1\right), 297.2575$ $\left(\mathrm{C}_{22} \mathrm{H}_{33}, 2\right), 285.2535\left(\mathrm{C}_{21} \mathrm{H}_{33}, \mathrm{~A}-\mathrm{Me}, 3\right), 283.2382\left(\mathrm{C}_{21} \mathrm{H}_{31}\right.$, 3), $255.2129\left(\mathrm{C}_{19} \mathrm{H}_{27}, 3\right), 241.1982\left(\mathrm{C}_{17} \mathrm{H}_{25}, 4\right)$ and 43 (100). Alkaline hydrolysis of acetate 2 c yielded alcohol $\mathbf{1 c}, \mathrm{mp} 165-$ $167^{\circ} \mathrm{C}, \mathrm{R} \mathrm{t}_{\mathrm{R}}(\mathrm{II})(\mathrm{GLC}) 1.95 ; m / z 440.3992\left(\mathrm{C}_{31} \mathrm{H}_{52} \mathrm{O}, \mathrm{M}^{+}, 11 \%\right.$;
requires $M, 440.4015), 425.3734\left(\mathrm{C}_{30} \mathrm{H}_{49} \mathrm{O}, 27\right), 407.3651$ $\left(\mathrm{C}_{30} \mathrm{H}_{47}, 5\right), 393.3447\left(\mathrm{C}_{29} \mathrm{H}_{45} \mathrm{O}, 1\right), 341.2878\left(\mathrm{C}_{24} \mathrm{H}_{37} \mathrm{O}, 3\right)$, $323.2727\left(\mathrm{C}_{24} \mathrm{H}_{35}, 2\right), 300.2805\left(\mathrm{C}_{22} \mathrm{H}_{36}, A, 4\right), 285.2525$ $\left(\mathrm{C}_{21} \mathrm{H}_{33}, \mathrm{~A}-\mathrm{Me}, 4\right), 273.2289\left(\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{O}, 3\right)$, 259.2089 $\left(\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{O}, 7\right), 255.2132\left(\mathrm{C}_{19} \mathrm{H}_{27}, 3\right), 241.1983\left(\mathrm{C}_{18} \mathrm{H}_{25}, 6\right)$, $229.2002\left(\mathrm{C}_{17} \mathrm{H}_{25}, 5\right)$ and $55.0543\left(\mathrm{C}_{4} \mathrm{H}_{7}, 100\right)$.
(24S)-24-Methyltirucalla-5,25-dien-3 3 -yl acetate 2 d and (24S)-24-methyltirucalla-5,25-dien-3p-ol 1d. Compound 2d: $m / z$ $482.4106\left(\mathrm{C}_{33} \mathrm{H}_{54} \mathrm{O}_{2}, \mathrm{M}^{+}, 17 \%\right.$; requires $M$, 482.4121), $467.3867\left(\mathrm{C}_{32} \mathrm{H}_{51} \mathrm{O}_{2}, 27\right), 422.3847\left(\mathrm{C}_{31} \mathrm{H}_{50}, 2\right), 407.3647$ $\left(\mathrm{C}_{30} \mathrm{H}_{47}, 18\right), 397.3067\left(\mathrm{C}_{27} \mathrm{H}_{41} \mathrm{O}_{2}, 1\right), 357.2745\left(\mathrm{C}_{24} \mathrm{H}_{37} \mathrm{O}_{2}\right.$, 1), $337.2898\left(\mathrm{C}_{24} \mathrm{H}_{37}, 3\right)$, $315.2388\left(\mathrm{C}_{21} \mathrm{H}_{31} \mathrm{O}_{2}, 2\right), 301.2171$ $\left(\mathrm{C}_{20} \mathrm{H}_{29} \mathrm{O}_{2}, 4\right), 300.2716\left(\mathrm{C}_{22} \mathrm{H}_{36}, A, 3\right), 297.2536\left(\mathrm{C}_{22} \mathrm{H}_{33}, 3\right)$, $285.2598\left(\mathrm{C}_{21} \mathrm{H}_{33}, A-\mathrm{Me}, 5\right), \quad 283.2437\left(\mathrm{C}_{21} \mathrm{H}_{31}, 12\right)$, $255.2063\left(\mathrm{C}_{19} \mathrm{H}_{27}, 5\right), 241.1992\left(\mathrm{C}_{17} \mathrm{H}_{25}, 5\right), 229.1971\left(\mathrm{C}_{17} \mathrm{H}_{25}\right.$, 7) and 43 (100). Alkaline hydrolysis of acetate 2 d yielded alcohol 1d, $\mathrm{mp} 146-148^{\circ} \mathrm{C}, \mathrm{R} t_{\mathrm{R}}$ (II) (GLC) $1.94 ; m / z 440.4011$ $\left(\mathrm{C}_{31} \mathrm{H}_{52} \mathrm{O}, \mathrm{M}^{+}, 8 \%\right.$; requires $\left.M, 440.4015\right), 425.3786$ $\left(\mathrm{C}_{30} \mathrm{H}_{49} \mathrm{O}, 17\right), 407.3630\left(\mathrm{C}_{30} \mathrm{H}_{47}, 4\right), 355.2987\left(\mathrm{C}_{25} \mathrm{H}_{39} \mathrm{O}, 1\right)$, $341.2810\left(\mathrm{C}_{24} \mathrm{H}_{37} \mathrm{O}, 2\right), 311.2328\left(\mathrm{C}_{22} \mathrm{H}_{31} \mathrm{O}, 3\right), 300.2789$ $\left(\mathrm{C}_{22} \mathrm{H}_{36}, \mathrm{~A}, 4\right), 285.2552\left(\mathrm{C}_{21} \mathrm{H}_{33}, \mathrm{~A}-\mathrm{Me}, 4\right), 283.2390$ $\left(\mathrm{C}_{21} \mathrm{H}_{31}, 6\right), 273.2315\left(\mathrm{C}_{19} \mathrm{H}_{29} \mathrm{O}, 2\right)$, $271.2217\left(\mathrm{C}_{19} \mathrm{H}_{27} \mathrm{O}, 2\right)$, $259.2106\left(\mathrm{C}_{18} \mathrm{H}_{27} \mathrm{O}, 5\right), 255.2132\left(\mathrm{C}_{19} \mathrm{H}_{27}, 2\right), 241.1983$ $\left(\mathrm{C}_{18} \mathrm{H}_{25}, 4\right), 229.2005\left(\mathrm{C}_{17} \mathrm{H}_{25}\right)$ and $41.0385\left(\mathrm{C}_{3} \mathrm{H}_{5}, 100\right)$.
4,4-Dimethylcholest-5-en-3 $\mathbf{\beta}$-yl acetate 5 . This compound showed $m / z 456.3941\left(\mathrm{C}_{31} \mathrm{H}_{52} \mathrm{O}_{2}, \mathrm{M}^{+}, 15 \%\right.$; Calc. for $M$, 456.3963); $441.3706\left(\mathrm{C}_{30} \mathrm{H}_{49} \mathrm{O}_{2}, 1\right), 396.3720\left(\mathrm{C}_{29} \mathrm{H}_{48}, 23\right)$, $381.3523\left(\mathrm{C}_{28} \mathrm{H}_{45}, 38\right)$, $353.3159\left(\mathrm{C}_{26} \mathrm{H}_{41}, 3\right), 328.3101$ $\left(\mathrm{C}_{24} \mathrm{H}_{40}, 12\right), 313.2898\left(\mathrm{C}_{23} \mathrm{H}_{37}, 2\right), 283.2450\left(\mathrm{C}_{21} \mathrm{H}_{31}, 6\right)$, $274.2621\left(\mathrm{C}_{20} \mathrm{H}_{34}, A, 4\right), 259.2421\left(\mathrm{C}_{19} \mathrm{H}_{31}, A-\mathrm{Me}, 1\right)$, $247.2405\left(\mathrm{C}_{18} \mathrm{H}_{31}, 5\right), 241.1954\left(\mathrm{C}_{18} \mathrm{H}_{25}, 6\right), 227.1844\left(\mathrm{C}_{17} \mathrm{H}_{23}\right.$, 2), $215.1828\left(\mathrm{C}_{16} \mathrm{H}_{23}, 4\right)$ and $43(100)$; $\delta_{\mathrm{C}}$ and $\delta_{\mathrm{H}}$ : $\mathrm{C}-1$ [36.3; $1.18(\alpha), 1.74(\beta)], \mathrm{C}-2[23.9 ; 1.73(\beta), 1.83(\alpha)], \mathrm{C}-3[79.5 ; 4.49$, dd, $J 4.4$ and 11.7], C-4 [40.4], C-5 [149.0], C-6 [120.7; 5.56, dd, $J 2.9$ and 4.4], C-7 [32.5; 1.64( $\alpha$ ), 2.09( $\beta$ )], C-8 [30.9; 1.49], C-9 [50.8; 0.92], C-10 [36.7], C-11 [20.6; $1.36(\beta), 1.47(\alpha)], \mathrm{C}-12$ [39.7; 1.15( $\alpha$ ), 2.00( $\beta$ )], C-13 [42.2], C-14 [57.2; 0.94], C-15 [23.8; 1.17( $), 1.33(\beta)]$, C-16 [28.3; $1.26(\beta), 1.84(\alpha)], \mathrm{C}-17$ [56.0; 1.06], C-18 [11.9; 0.67, s], C-19 [21.4; 1.10, s], C-20 [35.8; 1.37], C-21 [18.7; 0.91, d, J 6.2], C-22 [36.2; 1.00, 1.34]; C-23 [24.2; 1.07, 1.59], C-24 [39.5; 1.11, 1.11], C-25 [28.0; 1.52], C-26 [22.6; 0.86, d, J 6.6], C-27 [22.8; 0.87, d, J 6.6], C-28 [27.2; 1.02, s], C-29 [25.0; 1.14, s], C-3' (COMe) [170.8] and C-3" (COMe) [21.4; 2.06, s].
5a-Lanost-7-en-3p-yl acetate 6f. $\delta_{\mathrm{C}}$ and $\delta_{\mathrm{H}}$ : C-1 [37.7; 1.28( $\alpha$ ), $1.79(\beta)]$, C-2 [24.0; 1.64, 1.64], C-3 [81.2; 4.52, dd-like, J 4.7 and 10.7], C-4 [37.5], C-5 [50.3; 1.22], C-6 [22.8; 1.91( $\beta$ ), 2.02() )], C-7 [116.4; 5.20, br d, J 5.5], C-8 [145.1], C-9 [47.1; 2.01], C-10 [35.5], C-11 [20.0; 1.62( $\beta$ ), $1.48(\alpha)$ ], C-12 [32.1; $1.65,1.65], \mathrm{C}-13$ [44.3], C-14 [52.0], C-15 [32.2; 1.23( $\beta$ ), $1.57(\alpha)], \mathrm{C}-16[27.6 ; 1.29(\alpha), 1.95(\beta)], \mathrm{C}-17$ [50.8; 1.49], C-18 [16.0; 0.64, s], C-19 [14.2; 0.89, s], C-20 [36.5; 1.35], C-21 [19.0; 0.88, d, J 6.6], C-22 [36.5; 1.00, 1.15], C-23 [24.1; 1.18, 1.32], C-24 [39.5; 1.13, 1.13], C-25 [28.0; 1.52], C-26 [22.6; 0.86 , d, J6.6], C-27 [22.8; 0.87, d, J6.6], C-28 [28.2; 0.87, s], C-29 [16.6; 0.96, s], C-30 [24.8; 0.97, d, J 0.7], C-3' (COMe) [171.1] and C-3" (COMe) [21.3; 2.05, s].

## References

1 The Staff of the L. H. Bailey Hortorium, Cornell University, Hortus Third, Macmillan, New York, NY, 1976, p. 186.
2 D. Lavie and E. Glotter, Fortschr. Chem. Org. Naturst., 1971, 29, 307.

3 G. Biglino, L. Cattel, O. Caputo and G. Nobili, Gazz. Chim. Ital., 1969, 99, 830.
4 S. Tanaka, C. Uno, M. Akimoto, M. Tabata, C. Honda and W. Kamisako, Planta Med., 1991, 57, 527.

5 P. J. Hylands and M. T. Oskoui, Phytochemistry, 1979, 18, 1843; P. J. Hylands, E. S. Mansour and M. T. Oskoui, J. Chem. Soc., Perkin Trans. I, 1980, 2933.

6 L. Cattel, G. Balliano and O. Caputo, Phytochemistry, 1979, 18, 861.

7 (a) T. Akihisa, T. Tamura, T. Matsumoto, D. S. Eggleston, W. C. M. C. Kokke and N. Shimizu, J. Chem. Soc., Perkin Trans. I, 1988, 439; (b) T. Akihisa, W. C. M. C. Kokke, T. Tamura and T. Nambara, Chem. Pharm. Bull., 1992, 40, 1199; (c) T. Akihisa, W. C. M. C. Kokke, J. A. Krause, D. S. Eggleston, S. Katayama, Y. Kimura and T. Tamura, Chem. Pharm. Bull., 1992, 40, 3280; (d) T. Akihisa, W. C. M. C. Kokke, Y. Kimura and T. Tamura, J. Org. Chem., 1993, 58, 1959; (e) T. Akihisa, K. Yasukawa, Y. Kimura, M. Takido, W. C. M. C. Kokke and T. Tamura, Chem. Pharm. Bull., 1994, 42, 1101; (f) T. Akihisa, K. Yasukawa, M. Takido, W. C. M. C. Kokke and T. Tamura, Phytochemistry, 1994, 36, 153.
8 A. Rahier and P. Benveniste, Analysis of Sterols and Biologically Significant Steroids, ed. W. D. Nes and E. J. Parish, Academic, New York, 1989, pp. 223-250.
9 (a) T. Itoh, T. Tamura, T. M. Jeong, T. Tamura and T. Matsumoto, Lipids, 1980, 15, 122; (b) M. B. Gewali, M. Hattori, Y. Tezuka, T. Kikuchi and T. Namba, Phytochemistry, 1990, 29, 1625; (c) W. D. Nes, R. Y. Wong, M. Benson and T. Akihisa, J. Chem. Soc., Chem. Commun., 1991, 1272.
10 C. B. Gamalath, A. A. L. Gunatilaka and S. Subramaniam, J. Chem. Soc., Perkin Trans. 1, 1989, 2259.
11 M. J. U. Ferreira, A. M. Lobo, C. A. O'Mahoney, D. J. Williams and H. Wyler, J. Chem. Soc., Perkin Trans. I, 1990, 185.
12 T. Akihisa, K. Yamamoto, T. Tamura, Y. Kimura, T. Iida T. Nambara and F. C. Chang, Chem. Pharm. Bull., 1992, 40, 789.

13 H. Ageta, K. Shiojima, Y. Arai, H. Suzuki and T. Kiyotani, Chem. Pharm. Bull., 1994, 42, 39.
14 M. Tsuda and G. J. Schroepfer, Jr., Chem. Phys. Lipids, 1979, 25, 49.
15 S. G. Wyllie, B. Amos and L. Toekes, J. Org. Chem., 1977, 42, 725.
16 H. Budzikiewicz, J. M. Wilson and C. Djerassi, J. Am. Chem. Soc., 1963, 85, 3688; H. E. Audier and B. C. Das, Tetrahedron Lett., 1966, 2205.

17 CAChe with extended MM2 parameters (CAChe Scientific Inc., Beaverton, Oregon, USA). The conformation with minimum steric energy was obtained from the potential-energy map using the 'Sequential Search' option. Drawings were performed by the Chem3D program (Cambridge Scientific Computing Inc., Cambridge, Massachusetts, USA).
18 W. D. Nes, R. Y. Wong, M. Benson, J. R. Landrey and W. R. Nes, Proc. Natl. Acad. Sci. USA, 1984, 81, 5896.
19 I. Rubinstein, L. J. Goad, A. D. H. Clague and L. J. Mulheirn, Phytochemistry, 1976, 15, 195; T. Akihisa, N. Shimizu, R. Kawaguchi, T. Tamura and T. Matsumoto, J. Jpn. Oil Chem. Soc., 1986, 35, 907. Although the reference compounds used for comparison were cholestane-, lanostane- and cycloartane-type compounds, which possess a $20 R$ chirality, the stereochemistry at C-20 as well as the skeletal structure exerts almost no influence on the shielding of ${ }^{1} \mathrm{H}$ signals from the terminal protons of the sidechain as is also evident from the ${ }^{1} \mathrm{H}$ NMR data of compounds 4 b and 4 e (Table 2)
20 Dictionary of Terpenoids, ed. J. D. Connolly and R. A. Hill, Chapman and Hall, London, 1991, vol. 2.
21 C. R. Narayanan and D. G. Naik, Indian J. Chem., Sect. B, 1982, 20 , 62.

22 A. P. Dantanarayana, N. S. Kumar and M. U. S. Sultanbawa, J. Chem. Soc., Perkin Trans. 1, 1981, 2717.

23 L. K. Yankov and S. M. Husein, C. R. Acad. Bulg. Sci., 1975, 28, 1641. The structural assignment of 'citrullonol' should be reinvestigated because these authors reported a C-6 vinyl proton at $\delta 5.24$ ( ${ }^{1} \mathrm{H}$ NMR; $\mathrm{CDCl}_{3}$ ). However, in $\Delta^{5}$-unsaturated triterpenes the C-6 vinyl proton is known to resonate at $\delta 5.5-5.6 .^{9-13}$
24 W. R. Nes and M. L. McKean, Biochemistry of Steroids and Other Isopentenoids, University Park Press, Baltimore, Maryland, 1977; P. Benveniste, Annu. Rev. Plant Physiol., 1986, 37, 275.

25 T. Itoh, T. Tamura and T. Matsumoto, Lipids, 1976, 11, 434 T. Akihisa, Y. Kimura, K. Roy, P. Ghosh, S. Thakur and T. Tamura, Phytochemistry, 1994, 35, 1309.
26 T. Itoh, T. Tamura, T. Iida and T. Matsumoto, Steroids, 1974, 23 687.

27 N. Shimizu, T. Itoh, M. Saito and T. Matsumoto, J. Org. Chem., 1984, 49, 709.
28 T. Akihisa, Y. Kimura, W. C. M. C. Kokke, T. Itoh and T. Tamura, Chem. Pharm. Bull., 1996, 44, 1202.

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[^0]:    $\dagger$ We chose to use the trivial names of most compounds because consistent use of systematic names would have confused the reader. Tirucallane $=(13 \alpha, 14 \beta, 17 \alpha, 20 S)$-lanostane; euphane $=(13 \alpha, 14 \beta, 17 \alpha)-$ lanostane and cycloartane $=9,19$-cyclo- $9 \beta$-lanostane .

[^1]:    ${ }^{a}$ Cholesteryl acetate has $\mathrm{R} t_{\mathrm{R}}=1.00 .^{b}$ In each group, $R_{\mathrm{f}}$ of the triterpene acetates with a $\mathrm{C}_{\mathbf{8}}$-saturated side-chain ( $\mathbf{7 f}$ or $\mathbf{2 a}$ ) $=1.00$.

